Instructors:

Dr. Kyle Beran
   Email: Kyle.Beran@angelo.edu
   Phone: 486-6663
   Office: CAV 102B
   Office Hours: MWF 2-3, TR 11-12, or by appointment

Mr. Kevin Boudreaux
   Email: Kevin.Boudreaux@angelo.edu
   Phone: 486-6623
   Office: CAV 207B
   Office Hours: MWF 10-11, TR 11-12, or by appointment; Review sessions W 5 pm

Dr. David Carter
   Email: David.Carter@angelo.edu
   Phone: 486-6626
   Office: CAV 218
   Office Hours: M 1:30-3 pm; W 8-9:30, R 9:30-11:30 or by appointment

Dr. Edith Osborne
   Email: Edith.Osborne@angelo.edu
   Phone: 486-6629
   Office: CAV 204A
   Office Hours: All office hours will be held virtually. Please email to make an appointment.

Dr. Gregory Smith
   Email: Gregory.Smith@angelo.edu
   Phone: 486-6628
   Office: CAV 207A
   Office Hours: MTWRF 9-10, or by appointment

Dr. Ralph Zehnder
   Email: Ralph.Zehnder@angelo.edu
   Phone: 486-6662
   Office: CAV 204B
   Office Hours: W 2:30-5, F 11-1:30, or by appointment
Laboratory Meeting Times
The CHEM 1111 lab classes that accompany the CHEM 1311 lecture course are shown in the table below.

<table>
<thead>
<tr>
<th>Section</th>
<th>Day</th>
<th>Meeting Time</th>
<th>Instructor</th>
<th>Class Room</th>
<th>Lab Room</th>
</tr>
</thead>
<tbody>
<tr>
<td>08Z</td>
<td>M</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Zehnder</td>
<td>CAV 219</td>
<td>CAV 216</td>
</tr>
<tr>
<td>09Z</td>
<td>M</td>
<td>11:00 pm-01:50 pm</td>
<td>Dr. Osborne</td>
<td>CAV 215</td>
<td>CAV 216</td>
</tr>
<tr>
<td>10Z</td>
<td>M</td>
<td>02:00 pm-04:50 pm</td>
<td>Dr. Smith</td>
<td>CAV 211</td>
<td>CAV 212</td>
</tr>
<tr>
<td>11Z</td>
<td>T</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Carter</td>
<td>CAV 211</td>
<td>CAV 212</td>
</tr>
<tr>
<td>12Z</td>
<td>T</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Osborne</td>
<td>CAV 215</td>
<td>CAV 216</td>
</tr>
<tr>
<td>13Z</td>
<td>W</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Zehnder</td>
<td>CAV 215</td>
<td>CAV 216</td>
</tr>
<tr>
<td>14Z</td>
<td>W</td>
<td>02:00 pm-04:50 pm</td>
<td>Dr. Smith</td>
<td>CAV 211</td>
<td>CAV 212</td>
</tr>
<tr>
<td>15Z</td>
<td>W</td>
<td>02:00 pm-04:50 pm</td>
<td>Dr. Carter</td>
<td>CAV 215</td>
<td>CAV 216</td>
</tr>
<tr>
<td>16Z</td>
<td>R</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Zehnder</td>
<td>CAV 211</td>
<td>CAV 212</td>
</tr>
<tr>
<td>17Z</td>
<td>R</td>
<td>02:00 pm-04:50 pm</td>
<td>Dr. Boudreaux</td>
<td>CAV 215</td>
<td>CAV 212</td>
</tr>
<tr>
<td>18Z</td>
<td>R</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Osborne</td>
<td>CAV 215</td>
<td>CAV 216</td>
</tr>
</tbody>
</table>

The CHEM 1111 General Chemistry laboratory class accompanies the lecture class. The lab is designed to illustrate some of the principles involved in performing scientific measurements, handling chemicals, and performing chemistry experiments. In some cases, the experiments in the lab will introduce you to concepts before you cover them in the lecture course, and in some cases, the experiments will reinforce concepts already covered in the lecture course.

Required Texts and Materials for Laboratory
- **Textbook:**

Online virtual labs:
LearnSmart Labs – McGraw-Hill – FREE with this code CT0I-XBU8-6D3K-7XK2-2OYZ

- **Respondus Lockdown & Respondus Monitor** Access through Blackboard. Make sure that your computer/laptop is compatible with Respondus software. Respondus Monitor requires a webcam. Chromebooks are not compatible with Respondus. Lockdown and Monitor will be used for the administration of exams.

- **Approved Lab Goggles [Required]** (available from the ASU Bookstore or from the lab stockroom)

- **Calculator [Required]:** Scientific calculator capable of performing calculations with scientific notation and logarithms. *Bring your calculator to class and to lab every day. Only non-programmable calculators may be used on the exams.*

Course Description
1311/CHEM 1311 General Chemistry I (3-0). An introduction to the fundamental laws and theories of chemistry, chemical nomenclature, stoichiometry, atomic structure, chemical bonding, periodic table, chemical equations and reactions, and the properties
of heat flow and gases. **Prerequisites:** Students must have received: a score of 580 or above on the mathematics portion of the SAT if taken before March 2016, a score of 600 or above on the mathematics portion of the SAT if taken in March 2016 or after, a score of 26 or above on the mathematics section of the ACT, completed college algebra with a grade of “C” or better, or completed Chemistry 1305 with a grade of “C” or better in order to enroll in Chemistry 1311/1111. **Corequisite:** Chemistry 1111. 

1111/CHEM 1111 General Chemistry I Laboratory (0-3). Laboratory experiments that focus on laboratory technique, data collection, and analysis. The experiments will expand upon the concepts and topics presented in Chemistry 1311. **Corequisite:** Chemistry 1311.

**Course Delivery**
To maintain academic quality while accommodating physical distancing needs this semester, this course will use a split delivery model that combines face-to-face teaching with remote instruction.

**How Does It Work?**
Your lab sections will be divided, and you will be placed into two laboratory rooms (Cav 212 and Cav 216) to maintain physical distancing requirements in our assigned laboratory space.
You will perform laboratory experiments face to face or online alternating weekly.
Please pay close attention to the laboratory schedule, emails, and announcements posted on Blackboard as things will likely be changing regularly.
Please refer to this Health and Safety web page for updated information about campus guidelines as they relate to the COVID-19 pandemic.

**Technology Requirements**
**LearnSmart Labs from McGraw-Hill**
In order to access the LearnSmart Labs from McGraw-Hill, you will have to register through their connect website.
The weblink and the exact instructions for registering with your individual section is posted in your BlackBoard section.
You will need to enter the following code to gain free access:

CTOI-XBU8-6D3K-7XK2-2OYZ

**DO NOT** pay for the virtual lab registration as they are provided for free.

**Respondus Lockdown Browser**
Access to exams and quizzes will be through Respondus Lockdown Browser and will be video recorded via Respondus Monitor. Respondus requires a desktop computer or laptop (not a Chromebook) and a webcam. For best results, use an ethernet cable to connect to your Internet source instead of relying on Wifi. Refer to the Blackboard course for Respondus installation instructions.
Click here for more information:
https://www.youtube.com/watch?time_continue=9&v=XuX8WoeAycs&feature=emb_title
Attendance
You are expected to attend all face to face lab meetings according to the schedule of hands on and virtual laboratory experiments. You will not be automatically dropped if you stop attending class.

If you feel sick, please stay home. Keep your professor informed as to your status by email (preferred) or telephone (if necessary). Your faculty will work with you to keep up to date in the class.

Grading

Evaluation and Grades
Course grades will be determined as indicated in the table below.

<table>
<thead>
<tr>
<th>Assessment</th>
<th>Total Points</th>
</tr>
</thead>
<tbody>
<tr>
<td>Exams (3×100 pts)</td>
<td>300 pts</td>
</tr>
<tr>
<td>Final Exam</td>
<td>150 pts</td>
</tr>
<tr>
<td>Quizzes, classroom participation</td>
<td>150 pts</td>
</tr>
<tr>
<td>ALEKS Homework</td>
<td>200 pts</td>
</tr>
<tr>
<td>Laboratory</td>
<td>200 pts</td>
</tr>
<tr>
<td>Total</td>
<td>1000 pts</td>
</tr>
</tbody>
</table>

Students who are taking both CHEM 1111 and CHEM 1311 for the first time who wish to drop either course must drop both courses, because dropping either course would result in the co-requisite requirement no longer being met.

Overall grades in CHEM 1311/1111 will be determined as follows:

- If BOTH CHEM 1311 and CHEM 1111 are completed, the letter graded will be based on a total of 1000 points.
- For students who begin and complete ONLY CHEM 1311, a percentage will be calculated using only lecture assessments (first four items listed above with 800 points possible) and the letter grade will be assigned based on that percentage.
- For students who begin and complete ONLY CHEM 1111, the percentage will be calculated using only lab assessments and that percentage will be used to assign a letter grade.

Grading System
Course grades will be dependent upon completing course requirements and meeting the student learning outcomes.

The following grading scale will be used for this course:

A = 900-1000 points (90-100%)
B = 800-899 points 80-89.9%)
C = 700-799 points (70-79.9%)
D = 600-699 points (60-69.9%)
Blackboard
Grades will be posted on Blackboard. Information, handouts, homework assignments, and other course documents will either be posted on your instructor's faculty web page, or on Blackboard.

Last Day to Drop
The last day to drop the course with a grade of “W” is Tuesday, November 10, 2020.

Laboratory Attire
Beginning on the first day of lab, everyone MUST have approved goggles, long-sleeved shirts which cover the midriff, long pants, and shoes with closed toes and heels (no sandals, slides, etc.). (Basically, you should have as little exposed skin as possible.) Anyone not wearing the appropriate attire will not be allowed into lab.

Lab Procedures and Lab Reports
The procedures for face to face labs will be posted on Blackboard in the section labeled “Lab Resources.” The procedures will provide a description of the background for each experiment, pre-laboratory questions that will be checked off at the beginning of the lab period, a procedure for the experiment, and a lab report form which must be submitted on Blackboard. It is essential that you read the materials posted in Blackboard for that week’s lab before coming to lab. Virtual labs will be found on LearnSmart, and they must be completed by their due date. Each lab will be worth 100 points.

Cleaning Up After Lab
Make sure that your lab area is clean and that all glassware and hardware has been cleaned and returned to the appropriate places before leaving the lab.

Make-Up Lab Policy
The lowest lab score will be dropped from the total. If you miss a lab for a valid reason, that is the score that will be dropped.

“Lab Safety and Chemical Hygiene” Training
All students enrolled in lab courses are required to take a Mandatory “Lab Safety and Chemical Hygiene” course and Quiz on Blackboard. Instructions for completing the quiz are given below:
1. Login to Blackboard, and choose the course: entitled “Lab Safety and Chemical Hygiene.”
2. Under the left-hand menu, choose: “Get Started Here”.
3. There are three sections:
   a. Welcome to Lab Safety and Chemical Hygiene Training! — There are your instructions.
   b. Lab Safety and Chemical Hygiene Training — Click on “CHP Lab Safety Training 2018”. This will download a PowerPoint slide show which will cover the safety training. After viewing, remember to click the “Reviewed” button.
c. Lab Safety Chem Hygiene Quiz. You must score 90% or higher. You can take it again in 24 hours.

The Lab Safety Training must be completed by the evening of Thursday, August 27.

**Lab Final Exam**

There will be a 100-point lab final exam given during the last week of classes. This will be taken online using Respondus lockdown browser and monitoring with webcam. This grade will not be dropped from the lab total.

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**CLASS AND LAB SCHEDULE — FALL, 2020**

<table>
<thead>
<tr>
<th>Week</th>
<th>Week of</th>
<th>Lab Group B</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>8/17</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Lab Safety Lecture</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Significant figure lecture and worksheet</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Mandatory Lab Safety Training and Quiz — instructions given in <em>Lab Safety Training</em> section (must be completed by Sep. 8)</td>
</tr>
<tr>
<td>2</td>
<td>8/24</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Density</td>
</tr>
<tr>
<td>3</td>
<td>8/31</td>
<td>FACE TO FACE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Measuring mass and volume</td>
</tr>
<tr>
<td>4</td>
<td>9/07</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Take Home Assignment: “Scientific Measurement and Presentation of Data” must be completed before start of lab next week</td>
</tr>
<tr>
<td>5</td>
<td>9/14</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Calorimetry</td>
</tr>
<tr>
<td>6</td>
<td>9/21</td>
<td>FACE TO FACE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Percentage Water in a Hydrate</td>
</tr>
<tr>
<td>7</td>
<td>9/28</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Stoichiometry</td>
</tr>
<tr>
<td>8</td>
<td>10/05</td>
<td>FACE TO FACE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Titration of Vinegar</td>
</tr>
<tr>
<td>9</td>
<td>10/12</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Reactions in Solution</td>
</tr>
<tr>
<td>10</td>
<td>10/19</td>
<td>FACE TO FACE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Hess’s Law</td>
</tr>
<tr>
<td>11</td>
<td>10/26</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Gas Law</td>
</tr>
<tr>
<td>12</td>
<td>11/02</td>
<td>FACE TO FACE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Emission Spectra of Atoms</td>
</tr>
<tr>
<td>13</td>
<td>11/09</td>
<td>ONLINE</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Molecular Structures and Shapes</td>
</tr>
<tr>
<td>14</td>
<td>11/16</td>
<td><strong>Lab Final Exam</strong></td>
</tr>
</tbody>
</table>
**General Policies Related to This Course**

All students are required to follow the policies and procedures presented in these documents:

- Angelo State University Student Handbook
- Angelo State University Catalog

**Academic Integrity**

Students are expected to maintain complete honesty and integrity in all work. Any student found guilty of any form of dishonesty in academic work is subject of disciplinary action and possible expulsion from ASU.

The College of Science and Engineering adheres to the university's Statement of Academic Integrity.

**Accommodations for Students with Disabilities**

ASU is committed to the principle that no qualified individual with a disability shall, on the basis of disability, be excluded from participation in or be denied the benefits of the services, programs or activities of the university, or be subjected to discrimination by the university, as provided by the Americans with Disabilities Act of 1990 (ADA), the Americans with Disabilities Act Amendments of 2008 (ADAAA) and subsequent legislation.

Student Disability Services is located in the Office of Student Affairs, and is the designated campus department charged with the responsibility of reviewing and authorizing requests for reasonable accommodations based on a disability. It is the student’s responsibility to initiate such a request by contacting an employee of the Office of Student Affairs, in the Houston Harte University Center, Room 112, or contacting the department via email at ADA@angelo.edu. For more information about the application process and requirements, visit the Student Disability Services website.

The employee charged with the responsibility of reviewing and authorizing accommodation requests is:

Dallas Swafford  
Director of Student Disability Services  
Office of Student Affairs  
325-942-2047  
dallas.swafford@angelo.edu  
Houston Harte University Center, Room 112
Communication
Faculty will respond to email and/or telephone messages within 24 hours during working hours Monday through Friday. Weekend messages may not be returned until Monday.

Written communication via email: All private communication will be done exclusively through your ASU email address. Check frequently for announcements and policy changes. In your emails to faculty, include the course name and section number in your subject line.

Virtual communication: Office hours and/or advising may be done with the assistance of the telephone, Collaborate, Skype, etc.

Incomplete Grade Policy
It is policy that incomplete grades be reserved for student illness or personal misfortune. Please contact faculty if you have serious illness or a personal misfortune that would keep you from completing course work. Documentation may be required. See ASU Operating Policy 10.11 Grading Procedures for more information.

Plagiarism
Plagiarism is a serious topic covered in ASU’s Academic Integrity policy in the Student Handbook. Plagiarism is the action or practice of taking someone else’s work, idea, etc., and passing it off as one’s own. Plagiarism is literary theft.

In your discussions and/or your papers, it is unacceptable to copy word-for-word without quotation marks and the source of the quotation. It is expected that you will summarize or paraphrase ideas giving appropriate credit to the source both in the body of your paper and the reference list.

Papers are subject to be evaluated for originality. Resources to help you understand this policy better are available at the ASU Writing Center.

Student Absence for Observance of Religious Holy Days
A student who intends to observe a religious holy day should make that intention known in writing to the instructor prior to the absence. See ASU Operating Policy 10.19 Student Absence for Observance of Religious Holy Day for more information.

Title IX Statement
Angelo State University is committed to providing and strengthening an educational, working, and living environment where students, faculty, staff, and visitors are free from sex discrimination of any kind. In accordance with Title VII, Title IX, the Violence Against Women Act (VAWA), the Campus Sexual Violence Elimination Act (SaVE), and other federal and state laws, the University prohibits discrimination based on sex, which includes pregnancy, and other types of Sexual Misconduct. Sexual Misconduct is a broad term encompassing all forms of gender-based harassment or discrimination and unwelcome behavior of a sexual nature. The term includes sexual harassment, nonconsensual sexual contact, nonconsensual sexual intercourse, sexual assault,
sexual exploitation, stalking, public indecency, interpersonal violence (domestic violence or dating violence), sexual violence, and any other misconduct based on sex. You are encouraged to report any incidents involving sexual misconduct to the Office of Title IX Compliance and the Director of Title IX Compliance/Title IX Coordinator, Michelle Boone, J.D. You may submit reports in the following manner:
Online: www.angelo.edu/incident-form
Face to Face: Mayer Administration Building, Room 210
Phone: 325-942-2022
Email: michelle.boone@angelo.edu
Note, as a faculty member at Angelo State, I am a mandatory reporter and must report incidents involving sexual misconduct to the Title IX Coordinator. Should you wish to speak to someone in confidence about an issue, you may contact the University Counseling Center (325-942-2371), the 24-Hour Crisis Helpline (325-486-6345), or the University Health Clinic (325-942-2171).
For more information about resources related to sexual misconduct, Title IX, or Angelo State’s policy please visit: www.angelo.edu/title-ix.

Required Use of Masks/Facial Coverings: As a member of the Texas Tech University System, Angelo State University has adopted the mandatory Facial Covering Policy to ensure a safe and healthy classroom experience. Current research on the COVID-19 virus suggests there is a significant reduction in the potential for transmission of the virus from person to person by wearing a mask/facial covering that covers the nose and mouth areas. Therefore, in compliance with the university policy students in this class are required to wear a mask/facial covering before, during, and after class. Faculty members may also ask you to display your daily screening badge as a prerequisite to enter the classroom. You are also asked to maintain safe distancing practices to the best of your ability. For the safety of everyone, any student not appropriately wearing a mask/facial covering will be asked to leave the classroom immediately. The student will be responsible to make up any missed class content or work. Continued non-compliance with the Texas Tech University System Policy may result in disciplinary action through the Office of Student Conduct.

Modifications to the Syllabus
This syllabus, including grade evaluation and course schedule, is subject to modification. In particular, the COVID-19 pandemic may require significant changes in course delivery and content on potentially short notice.

Student Learning Outcomes

• Learning Goal 1: Students will be able to analyze complex chemical problems and draw logical conclusions.
  o Students will be able to use an understanding of atomic structure at the basic and atomic levels to analyze the structure and reactivity of substances and chemical species.
• Students will be able to use an understanding of how energy interacts with matter to predict stable chemical species, and perform thermodynamic calculations describing chemical reactions.

**Learning Goal 2a:** Students will be able to understand and apply scientific reasoning in the chemical sciences.
- Students will be able to use an understanding of ions and molecules at the atomic level to predict the behavior of reactions in aqueous solutions.
- Students will be able to use the basic ideas of quantum mechanics to describe how molecular bonds form and to predict molecular shape and polarity. Molecular structure and polarity will be used to predict the forces between molecules and relate those forces to the states of matter and phase changes.

**Learning Goal 2b:** Students will be able to employ mathematics in the analysis of chemical problems.
- The mole concept, chemical formulas and balanced chemical equations will be used to do chemical calculations that relate macroscopic measurements to numbers of atoms, ions or molecules.
- Students will be able to do calculations involving solution concentration and know how to prepare solutions of given concentrations.
- Students will be able to quantitatively predict gas properties using gas law calculations.

**Learning Goal 3:** Students will be able to demonstrate technical and analytical skills in chemistry.
- Students will be able to use the periodic table to determine basic atomic information and to predict trends in atomic properties.
- Students will be able to interconvert between chemical names and formulas to the extent that they can work problems given only one of those pieces of information.
- Students will be able to classify common types of chemical reactions and predict the outcomes of reactions.

**Evaluation of Student Learning Outcomes**
Student learning outcomes will be evaluated by test questions or by the grading of in-classroom activities, as described by your instructor.

**Texas Higher Education Coordinating Board Natural Sciences Objectives**
The objective of the study of a natural sciences component of a core curriculum is to enable the student to understand, construct, and evaluate relationships in the natural sciences, and to enable the student to understand the basis for building and testing theories.

**Exemplary Educational Objectives**
1. To understand and apply method and appropriate technology to the study of natural sciences.
2. To recognize scientific and quantitative methods and the differences between these approaches and other methods of inquiry and to communicate findings, analyses, and interpretation both orally and in writing.
3. To identify and recognize the differences among competing scientific theories.
4. To demonstrate knowledge of the major issues and problems facing modern science, including issues that touch upon ethics, values, and public policies.
5. To demonstrate knowledge of the interdependence of science and technology and their influence on, and contribution to, modern culture.

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2. http://blackboard.angelo.edu (or access Blackboard from RamPort)
4. https://www.angelo.edu/catalogs/
6. https://www.angelo.edu/services/disability-services/
7. https://www.angelo.edu/content/files/14197-op-1011-grading-procedures
10. https://www.angelo.edu/content/files/14206-op-1019-student-absence-for-observance-of