Syllabus
CHEM 1111 General Chemistry Laboratory — Spring 2022
Department of Chemistry and Biochemistry, Angelo State University

Instructor:
Name : Dr. Shanmugapriya Dharmarajan
Office : CAV 204A
Office hours : M 10:00 am to 12:00 pm, T 12:30 to 01:30 pm, W 12:30 to 02:30 pm
E-mail : sdharmarajan@angelo.edu
Phone : 325-486-6626

Course Materials:
LabFlow – Catalyst Education (Required): $49.99 per semester. You would also need access to
LabFlow, a platform that you will use to upload your lab reports and other lab related assignments. For
signup instructions please follow the instructions given on our Blackboard course.

Respondus Lockdown Browser & Respondus Monitor (Required): In some occasions, exams and
quizzes may be administered through Respondus Lockdown Browser. It requires a desktop computer
or laptop and a webcam and a microphone. Make sure that your computer/laptop is compatible with
Respondus software. To download and install the latest version of the software, follow the instructions
given on our Blackboard course.

Chemistry Lab Goggles (Required): Available at ASU Bookstore or you can purchase from our general
chemistry lab stockroom on your first day of lab ($15).

Calculator (Required): A scientific calculator capable of performing calculations with scientific notation
and logarithms. Bring your calculator to class and to lab every day. Note: You can only use a non-
programmable calculator for the exam.

Course Description:
CHEM 1111 General Chemistry I Laboratory: Laboratory experiments that focus on laboratory
 technique, data collection, and analysis. The experiments will expand upon the concepts and topics
presented in Chemistry 1311. Corequisite: Chemistry 1311.

Laboratory Meeting Times:
The lab classes that accompany the CHEM 1311 lecture course are shown in the table below. The labs
will meet first for pre-lab lectures in the Class Room listed in the table.

<table>
<thead>
<tr>
<th>Section</th>
<th>Day</th>
<th>Meeting Time</th>
<th>Instructor</th>
<th>Class Room</th>
<th>Lab Room</th>
</tr>
</thead>
<tbody>
<tr>
<td>06Z</td>
<td>M</td>
<td>11:00 am-01:50 pm</td>
<td>Mr. Rilling</td>
<td>CAV 211</td>
<td>CAV 216</td>
</tr>
<tr>
<td>07Z</td>
<td>W</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Zehnder</td>
<td>CAV 211</td>
<td>CAV 216</td>
</tr>
<tr>
<td>08Z</td>
<td>T</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Zehnder</td>
<td>CAV 211</td>
<td>CAV 216</td>
</tr>
<tr>
<td>09Z</td>
<td>R</td>
<td>11:00 am-01:50 pm</td>
<td>Dr. Zehnder</td>
<td>CAV 211</td>
<td>CAV 216</td>
</tr>
<tr>
<td>10Z</td>
<td>M</td>
<td>02:00 pm-04:50 pm</td>
<td>Dr. Dharmarajan</td>
<td>CAV 211</td>
<td>CAV 216</td>
</tr>
<tr>
<td>11Z</td>
<td>T</td>
<td>02:00 pm-04:50 pm</td>
<td>Dr. Dharmarajan</td>
<td>CAV 223</td>
<td>CAV 216</td>
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</table>
The CHEM 1111 General Chemistry laboratory class accompanies this lecture class. The lab is designed to illustrate some of the principles involved in performing scientific measurements, handling chemicals, and performing chemistry experiments. In some cases, the experiments in the lab will introduce you to concepts before you cover them in the lecture course, and in some cases, the experiments will reinforce concepts already covered in the lecture course. Bring your calculator!

**Laboratory Attire:**
Beginning on the first day of lab, everyone MUST have approved goggles, long-sleeved shirts which cover the midriff, long pants, and shoes with closed toes and heels (no sandals, slides, etc.). (Basically, you should have as little exposed skin as possible.) Anyone not wearing the appropriate attire will not be allowed into lab.

**Lab Procedures and Lab Reports:**
We will be using the platform LabFlow by Catalyst Education for lab procedures and reports. You will have to register and to create an account with LabFlow by following instruction given on Blackboard. While you sign up you will be prompted to pay a fee of $49.99. The procedures for the labs will be posted within the LabFlow platform. The procedures will provide a description of the background for each experiment. You will be asked to complete pre-laboratory questions prior to the lab meeting within the LabFlow platform.

Your instructor will communicate to you at what time the prelab assignments are due. The lab report itself will be uploaded to LabFlow and must be turned in by whichever due date your instructor requires. Each lab will be worth up to 100 points (up to 20 points for the prelab questions, and up to 80 points for the lab report).

Labs will begin meeting in the second week of the semester. Bring your calculator!

During the first week there are a few dry activities you must complete online.

**Cleaning Up After Lab:**
Make sure that your lab area is clean and that all glassware and hardware has been cleaned and returned to the appropriate drawers before leaving the lab.

**Make-Up Lab Policy:**
If you have to miss a lab you will be turning in a lab report based on virtual data LabFlow will generate for you. You will have to seek permission from your instructor to be eligible to turn in a 100% virtual lab report. Your instructor will first work with you to attend another face-to-face lab section if that is feasible.

**Lab Safety Training:**
All students enrolled in lab courses are required to take a Mandatory Laboratory Safety Training and Quiz on Blackboard. Instructions for completing the quiz are given below:
1. Login to Blackboard, and choose the course entitled: “Lab Safety and Chemical Hygiene.”
2. Under the left-hand menu, choose: “Get Started Here”.
3. Click on “Get Started Here” in the left-hand column.
4. Follow the instructions under “Welcome to Lab Safety and Chemical Hygiene Training!”
5. You must score 90% or higher on the lab safety quiz.

The Lab Safety Training must be completed by the evening of **Sunday, January 23.**
Lab Midterm and Final Exams:
There will be a 100-point lab midterm (week of 3/7) and a 100-point lab final exam (week of 5/2).

### LAB SCHEDULE — Spring 2022

<table>
<thead>
<tr>
<th>Week #</th>
<th>Week Of</th>
<th>Lecture</th>
<th>Lab</th>
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</thead>
</table>
| 1      | 01/17   | Monday, Jan 17 - MLK day – No classes  
Chapter 1: Essential Ideas — Composition of Matter, States of Matter, Measurements, SI Units, Significant Figures, Unit Conversions | Labs do not meet  
Safety videos and Quiz, Chemistry Math and Labware videos and Quiz  
Mandatory Lab Safety Training and Quiz — instructions given in Lab Safety Training section |
| 2      | 01/24   | Chapter 2: Atoms, Ions, and Molecules — Nuclear Model, Atomic Mass, Periodic Table, Molecular and Ionic Compounds, Naming Compounds and Writing Formulas | Conversion Factors and Problem Solving (dry lab) |
| 3      | 01/31   | Chapter 2, cont.  
Quiz 1 – Thurs., Feb. 3 | Basic Laboratory Techniques |
| 4      | 02/07   | Chapter 3: Composition of Substances and Solutions — The Mole Concept, Empirical and Molecular Formulas, Molarity, Percent Composition | Density and Specific Gravity |
| 5      | 02/14   | Chapter 3, cont.  
Exam 1 – Wed., Feb. 16 (Ch. 1, 2,3) | Empirical Formulas |
| 6      | 02/21   | Chapter 4: Stoichiometry — Balancing Equations, Classifying Reactions, Stoichiometry, Limiting Reactants, Reaction Yield | Chemistry of Copper and Percent Yield |
| 7      | 02/28   | Chapter 4, cont.  
Quiz 2 – Thurs., Mar. 3 | Solutions, Electrolytes, and Concentrations |
| 8      | 03/07   | Chapter 5: Thermochemistry — Energy, Systems, Enthalpy, Heat Capacity, Calorimetry, Hess’s Law, Standard Enthalpies of Formation | Oxidation Reduction Reactions  
Lab Midterm Exam (covering the lab material from week 1 through week 7) |
| 9      | 03/14   | Spring Break – No classes | Spring Break – No Labs |
| 10     | 03/21   | Chapter 6: Electronic Structure and Periodicity — Electromagnetic Radiation, Bohr Model, Quantum Theory, Electronic Configurations, Periodic Trends of Elements  
Exam 2 – Wed., Mar. 23 (Ch. 3, 4, 5, 6) | Titration: Determining the Concentration of an Acid |
<p>| 11     | 03/28   | Chapter 6, cont. | Energy and Specific Heat |</p>
<table>
<thead>
<tr>
<th>Date</th>
<th>Day</th>
<th>Chapter/Section</th>
<th>Lecture Topic</th>
</tr>
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<tbody>
<tr>
<td>12</td>
<td>04/04</td>
<td>Chapter 7: Chemical Bonding and Molecular Geometry — Ionic and Covalent Bonding, Lewis Structures, Formal Charges/Resonance, Strengths of Bonds, VSEPR Theory, Structure and Polarity Quiz 3 – Thurs., Apr. 7</td>
<td>Constant Pressure Calorimetry</td>
</tr>
<tr>
<td>13</td>
<td>04/11</td>
<td>Chapter 8: Covalent Bonding — Valence Bond Theory, Hybrid Orbitals, Multiple Bonds, Molecular Orbital Theory Friday, April 15 – Spring holiday – No classes</td>
<td>Atomic Spectra</td>
</tr>
<tr>
<td>14</td>
<td>04/18</td>
<td>Chapter 9: Gases — Pressure, Temperature, Volume, Ideal Gas Law, Gas-Phase Stoichiometry, Effusion/Diffusion, Kinetic-Molecular Theory Exam 3 – Wed., Apr. 20 (Ch. 6, 7, 8, 9)</td>
<td>Modeling Geometry and Polarity</td>
</tr>
<tr>
<td>15</td>
<td>04/25</td>
<td>Thursday, April 28: Last Day to Drop Chapter 9, cont.</td>
<td>Analysis of KClO3/KCl using the Ideal Gas Law</td>
</tr>
<tr>
<td>16</td>
<td>05/02</td>
<td>Chapter 10: Intermolecular Forces — Intermolecular Forces, Vapor Pressure, Phase Diagrams, Water Quiz 4 – Thurs., May 5</td>
<td>Lab Final Exam</td>
</tr>
<tr>
<td>17</td>
<td>05/09</td>
<td>Final Exams</td>
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</table>

**Course and University Policies:**

All students are required to follow the policies and procedures presented in these documents:

- [Angelo State University Student Handbook](#)
- [Angelo State University Catalog](#)

**Office Hours / Email Communication:** Students can walk-in to the regular office hours. If you cannot make it during those hours, please make an appointment by email. Students are expected to frequently check their Angelo State email account and the Blackboard course website announcements for important communication from the instructor. Use CHEM1111 in the subject line of your emails to enable proper filtering. The instructor will only answer e-mails that are sent from an ASU e-mail account. The instructor will respond to legitimate e-mails within 24-48 hours during the week and may not respond until after weekends or holidays if messages are received on any of such days. More general questions will be addressed in the following lecture.

**Academic Integrity:** Students are expected to maintain complete honesty and integrity in all work. Any student found guilty of any form of dishonesty in academic work is subject of disciplinary action and possible expulsion from ASU.
The College of Science and Engineering adheres to the university’s Statement of Academic Integrity.  

**Accommodations for Students with Disabilities:** ASU is committed to the principle that no qualified individual with a disability shall, on the basis of disability, be excluded from participation in or be denied the benefits of the services, programs or activities of the university, or be subjected to discrimination by the university, as provided by the Americans with Disabilities Act of 1990 (ADA), the Americans with Disabilities Act Amendments of 2008 (ADAAA) and subsequent legislation.

Student Disability Services is located in the Office of Student Affairs, and is the designated campus department charged with the responsibility of reviewing and authorizing requests for reasonable accommodations based on a disability. It is the student’s responsibility to initiate such a request by contacting an employee of the Office of Student Affairs, in the Houston Harte University Center, Room 112, or contacting the department via email at ADA@angelo.edu. For more information about the application process and requirements, visit the Student Disability Services website.  

The employee charged with the responsibility of reviewing and authorizing accommodation requests is:

Dr. Dallas Swafford  
Director of Student Disability Services  
Office of Student Affairs  
325-942-2047  
dallas.swafford@angelo.edu  
Houston Harte University Center, Room 112

**Incomplete Grade Policy:** It is policy that incomplete grades be reserved for student illness or personal misfortune. Please contact faculty if you have serious illness or a personal misfortune that would keep you from completing course work. Documentation may be required. See ASU Operating Policy 10.11 Grading Procedures for more information.

**Plagiarism:** Plagiarism is a serious topic covered in ASU’s Academic Integrity Statement in the Student Handbook. Plagiarism is the action or practice of taking someone else’s work, idea, etc., and passing it off as one’s own. Plagiarism is literary theft.

In your discussions and/or your papers, it is unacceptable to copy word-for-word without quotation marks and the source of the quotation. It is expected that you will summarize or paraphrase ideas giving appropriate credit to the source both in the body of your paper and the reference list.

Papers are subject to be evaluated for originality. Resources to help you understand this policy better are available at the ASU Writing Center.

**Student Absence for Observance of Religious Holy Days:** A student who intends to observe a religious holy day should make that intention known in writing to the instructor prior to the absence.  See ASU Operating Policy 10.19 Student Absence for Observance of Religious Holy Day for more information.

**Title IX at Angelo State University:** Angelo State University is committed to providing and strengthening an educational, working, and living environment where students, faculty, staff, and
visitors are free from sex discrimination of any kind. In accordance with Title VII, Title IX, the Violence Against Women Act (VAWA), the Campus Sexual Violence Elimination Act (SaVE), and other federal and state laws, the University prohibits discrimination based on sex, which includes pregnancy, and other types of Sexual Misconduct. Sexual Misconduct is a broad term encompassing all forms of gender-based harassment or discrimination and unwelcome behavior of a sexual nature. The term includes sexual harassment, nonconsensual sexual contact, nonconsensual sexual intercourse, sexual assault, sexual exploitation, stalking, public indecency, interpersonal violence (domestic violence or dating violence), sexual violence, and any other misconduct based on sex.

You are encouraged to report any incidents involving sexual misconduct to the Office of Title IX Compliance and the Director of Title IX Compliance/Title IX Coordinator, Michelle Miller, J.D. You may submit reports in the following manner:

Online: Incident Reporting Form

Face to Face: Mayer Administration Building, Room 210

Phone: 325-942-2022

Email: michelle.miller@angelo.edu

Note, as a faculty member at Angelo State, I am a mandatory reporter and must report incidents involving sexual misconduct to the Title IX Coordinator. Should you wish to speak to someone in confidence about an issue, you may contact the University Counseling Center (325-942-2371), the 24-Hour Crisis Helpline (325-486-6345), or the University Health Clinic (325-942-2171). For more information about resources related to sexual misconduct, Title IX, or Angelo State’s policy please visit the Title IX website.

Information about COVID-19: Please refer to ASU’s COVID-19 (Coronavirus) Updates web page for current information about campus guidelines and safety standards as they relate to the COVID-19 pandemic.

Modifications to the Syllabus: This syllabus, including grade evaluation and course/lab schedule, is subject to modification on potentially short notice based on developing circumstances.

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1 https://www.angelo.edu/current-students/student-handbook/
2 https://www.angelo.edu/academics/catalog/
3 https://www.angelo.edu/live/files/27603-student-handbook-2020-21#page=96
4 https://www.angelo.edu/current-students/disability-services/
5 https://angelo.policystat.com/policy/10659448/latest/
7 https://www.angelo.edu/current-students/writing-center/academic_honesty.php
8 https://angelo.policystat.com/policy/10659368/latest/
9 https://www.angelo.edu/incident-form
10 https://www.angelo.edu/title-ix
11 https://www.angelo.edu/covid-19/